



## Chemical Laws and Problems; For Use in Colleges and Schools

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By John J Pilley

Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book \*\*\*\*\* Print on Demand \*\*\*\*\*.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1891 Excerpt: .of air which occupies 26 cc. at a temperature of 59F and under a pressure of 29 inches of mercury, is raised in temperature to 68F, and the pressure is increased to 50 inches of mercury; what is the new volume of gas? 171. What volume of gas measured at 28C and 748 mm. pressure can be obtained from 5 grams of Formic acid when heated with sulphuric? 172. A gas has its temperature raised to 50C from 15C; at 50C it measures 15 litres; what volume did it occupy at 150? 173. If 120 cc. of a gas measured at 35C, and under a pressure of 30 inches is raised in temperature to 46C, whilst the pressure is reduced to 25 inches; find its volume. 174. How much phosphorus is required to remove all the oxygen from 100 litres of air at 15C, and 740...



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